# Studies on the Constituents of Asclepiadaceae Plants. Part 67. ${ }^{1}$ Further Studies on Glycosides with a Novel Sugar Chain Containing a Pair of Optically Isomeric Sugars, D- and L-Cymarose, from Cynanchum africanum 

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Two new glycosides named cynafoside-C (3) and -D (4) were isolated from Cynanchum africanum $R$. Br. (Asclepiadaceae), and their structures were deduced on the basis of spectral and chemical evidence. The absolute configurations of cymarose, oleandrose, and digitoxose included in compounds (3) and (4) were each determined by h.p.l.c. analysis of their carbamoyl derivatives on a chiral column after acidic hydrolysis of the glycosides. A unique feature is that cynafoside-C (3), as well as cynafoside-A (1) and$B(2)$, contains a pair of optically isomeric sugars, D -and L -cymarose, in the sugar chain.

Cynanchum africanum R. BR. (Asclepiadaceaea) causes stock losses in South Africa owing to cynanchosis, an intoxication affecting the nervous and muscular systems. ${ }^{2}$ In a previous paper ${ }^{3}$ we reported the isolation and the structural elucidation of two glycosides, cynafoside-A (1) and -B (2), from the dried leaf and stem of this plant. In this paper we describe the structures of two additional glycosides named cynafoside-C (3) and -D (4). The absolute stereochemistry of the sugar moieties, cymarose, oleandrose, and digitoxose, was ascertained by using a chiral h.p.l.c. column, and was confirmed by comparison with that of authentic material. ${ }^{4-6}$
compound (3) afforded an aglycone (5) and a mixture of sugars. The mass spectrum of aglycone (5) gave a molecular ion peak at $m / z 512$. The ${ }^{1} \mathrm{H}$ n.m.r. spectrum ( 100 MHz ) showed four tertiary methyl proton signals at $\delta_{\mathrm{H}} 0.98,1.16,1.65$, and 2.07 , two vicinal methine proton signals bearing ester groups at $\delta_{\mathrm{H}}$ $4.95(1 \mathrm{H}, \mathrm{d}, J 9.8 \mathrm{~Hz})$ and $5.44(1 \mathrm{H}, \mathrm{t}, J 9.8 \mathrm{~Hz})$, and five aromatic proton signals at $\delta_{\mathrm{H}} 7.46-7.65(3 \mathrm{H}, \mathrm{m})$ and $8.06(2 \mathrm{H}$, dd, $J 8.3$ and 2.0 Hz ), which were characteristic of cynafogenin, ${ }^{3}$ the aglycone common to both glycosides (1) and (2). By comparing m.p.s, specific rotations, and $R_{\mathrm{F}}$-values on t.l.c. with those of an authentic sample, compound (5) was confirmed to be



Cynafoside - $A$ (1) $R=R^{1}$
Cynafoside-B (2) $R=R^{2}$
Cynafoside-C (3) $R=R^{3}$
Cynafoside-D (4) $R=R^{4}$
Cynafogenin (5) $R=H$

## Results and Discussion

Compound (3) has the molecular formula $\mathrm{C}_{63} \mathrm{H}_{96} \mathrm{O}_{24}$ on the basis of elemental analysis. The ${ }^{1} \mathrm{H}$ n.m.r. ( 500 MHz ) and the ${ }^{13} \mathrm{C}$ n.m.r. spectra (Tables 1 and 2) suggested that the structure of cynafoside-C (3) was analogous to those of (1) and (2). Five anomeric carbon signals at $\delta_{\mathrm{C}} 96.0,97.2,100.3,101.2$, and 102.1 revealed the presence of five sugars in compound (3). Acidic hydrolysis ( $0.025 \mathrm{M} \mathrm{H}_{2} \mathrm{SO}_{4}-75 \% \mathrm{MeOH} ; 60^{\circ} \mathrm{C} ; 15 \mathrm{~min}$ ) of
cynafogenin. The mixture of sugars was separated into four sugars (6), (7), (8), and (9) by column chromatography on silica gel. In the mass spectra of sugars (6), (7), and (8) molecular ion peaks were observed at $m / z 162,162$, and 148 , respectively, indicating monosaccharide units. In the ${ }^{1} \mathrm{H}$ n.m.r. spectrum $\left(\mathrm{CDCl}_{3}\right)$ of compound (6), four methoxy methyl groups appeared at $\delta_{\mathbf{H}} 3.31,3.37,3.44$, and 3.54 in the proportions $c a$. 15:10:9:7. The signal pattern was super-

Table 1. ${ }^{13} \mathrm{C}$ N.m.r. chemical shifts $\left(\delta_{\mathrm{C}}\right)$ of aglycone (5) and the aglycone moieties of glycosides (3), (4), and (10) in $\mathrm{C}_{5} \mathrm{D}_{5} \mathrm{~N}$

|  | (5) | (3) | (4) | (10) |
| :---: | :---: | :---: | :---: | :---: |
| C-1 | 39.7 | 37.9 | 38.1 | 38.1 |
| C-2 | 32.8 | $30.3(-2.5)$ | $30.4(-2.4)$ | $30.4(-2.4)$ |
| C-3 | 69.9 | 76.0 ( + 6.1) | 76.0 ( + 6.1) | $76.1(+6.2)$ |
| C-4 | 39.7 | 35.4 (-4.3) | 35.5 (-4.2) | 35.5 (-4.2) |
| C-5 | 45.1 | 44.6 | 44.6 | 44.7 |
| C-6 | $29.5{ }^{\text {a }}$ | $29.3{ }^{\text {a }}$ | $29.3{ }^{\text {a }}$ | $29.4{ }^{\text {a }}$ |
| C-7 | $28.5{ }^{\text {a }}$ | $28.4{ }^{\text {a }}$ | $28.4{ }^{\text {a }}$ | $28.4{ }^{\text {a }}$ |
| C-8 | 40.2 | 39.9 | 40.0 | 40.0 |
| C-9 | 50.2 | 50.1 | 50.1 | 50.1 |
| C-10 | 38.0 | 37.8 | 38.0 | 38.0 |
| C-11 | 71.7 | 71.7 | 71.6 | 71.6 |
| C-12 | 79.1 | 78.9 | $79.1{ }^{\text {b }}$ | 79.1 |
| C-13 | 55.0 | 54.8 | 54.9 | 55.0 |
| C-14 | 84.0 | 83.8 | 83.9 | 83.9 |
| C-15 | 33.9 | 33.9 | 33.9 | 33.9 |
| C-16 | 24.3 | 24.2 | 24.3 | 24.3 |
| C-17 | 58.5 | 58.3 | 58.4 | 58.4 |
| C-18 | 11.9 | 11.7 | 11.8 | 11.8 |
| C-19 | 12.5 | 12.3 | 12.4 | 12.4 |
| C-20 | 213.3 | 213.4 | 213.4 | 213.4 |
| C-21 | 31.7 | 31.7 | 31.7 | 31.7 |
| $\mathrm{MeCO}_{2}$ | 170.4 | 170.2 | 170.4 | 170.2 |
| $\mathrm{MeCO}_{2}$ | 21.4 | 21.3 | 21.3 | 21.3 |
| PhCO 2 | 166.8 | 166.8 | 166.7 | 166.7 |
| ipso- $\mathrm{PhCO}_{2}$ | 130.2 | 130.0 | 130.1 | 130.2 |
| $o-\mathrm{Ph} \mathrm{CO}_{2}$ | 130.2 | 130.0 | 130.1 | 130.2 |
| $m-\mathrm{PhCO} 2$ | 129.2 | 129.1 | 129.2 | 129.2 |
| $p-\mathrm{PhCO}_{2}$ | 133.9 | 133.9 | 133.9 | 133.9 |




Glaucobiose (9) L-Cymarose (11)
imposable on that of cymarose. The ${ }^{1} \mathrm{H}$ n.m.r. spectrum of compound (7) in $\mathrm{CDCl}_{3}$ was characteristic of oleandrose. The signals of $x$-and $\beta$-oleandropyranoses were observed in the ratio ca. 2:1. The ${ }^{1} \mathrm{H}$ n.m.r. spectrum $\left(\mathrm{CDCl}_{3}\right)$ of compound (8) revealed the signals due to $\beta$-digitoxopyranose. In $\mathrm{CDCl}_{3}$ digitoxose exists mostly as the $\beta$-digitoxopyranose form. The absolute configurations of compounds (6), (7), and (8) were each determined on the basis of h.p.l.c. analysis on a chiral column after conversion into methyl 4-O-(3,5-dinitrophenylcarbamoyl)glycosides for (6) ${ }^{4}$ and (7) ${ }^{5}$ and the 1-O-(3,5-dinitro-phenylcarbamoyl)-3,4-O-isopropylidene derivative for (8). ${ }^{6}$ The results of this h.p.l.c. analysis apparently demonstrated that these sugars all belonged to the D -series by direct comparison with authentic samples of their D - and L enantiomers (Figure 1). The chromatogram of the samples derived from compounds (6) and (7) showed four peaks. They correspond to carbamates of the four methyl glycosides

Table 2. ${ }^{13} \mathrm{C}$ N.m.r. chemical shifts $\left(\delta_{\mathrm{C}}\right)$ of the sugar moieties of glycosides (1), (2), (3), (4), and (10) in $\mathrm{C}_{5} \mathrm{D}_{5} \mathrm{~N}$

|  | (1) | (2) | (3) | (4) | (10) |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  | D-Cym | Digit | D-Cym | Digit | D-Cym |
| C-1 | 95.9 | 95.9 | 96.0 | 96.0 | 96.1 |
| C-2 | $37.1{ }^{\text {a }}$ | 39.0 | 37.1 | 39.0 | 37.3 |
| C-3 | $77.9{ }^{\text {b }}$ | 68.5 | 78.0 | $68.7^{\text {c }}$ | 78.1 |
| C-4 | $83.2^{\text {c }}$ | 83.3 | $83.2{ }^{\text {b }}$ | $83.4{ }^{\text {d }}$ | $83.4{ }^{\text {b }}$ |
| C-5 | $69.1{ }^{\text {d }}$ | 67.4 | 68.9 | $67.5^{\text {e }}$ | 69.1 |
| C-6 | $18.5{ }^{\text {e }}$ | $18.6{ }^{\text {a }}$ | $18.5^{\text {c }}$ | $18.7{ }^{\text {f }}$ | $18.6{ }^{\text {c }}$ |
| 3-OMe | $58.7{ }^{f}$ |  | $58.8{ }^{\text {d }}$ |  | $58.9{ }^{\text {d }}$ |
|  | D-Cym | D-Cym | Digit | Digit | Digit |
| C-1 | 100.2 | 99.6 | 100.3 | 99.8 | 100.5 |
| C-2 | $36.9^{\text {a }}$ | 36.8 | 38.6 | $38.7^{\text {c }}$ | 39.0 |
| C-3 | $77.6{ }^{\text {b }}$ | 77.6 | 68.3 | 68.7 | 68.5 |
| C-4 | $82.1{ }^{\text {c }}$ | 82.1 | $83.1{ }^{\text {b }}$ | $83.2{ }^{\text {d }}$ | $83.3{ }^{\text {b }}$ |
| C-5 | $68.9{ }^{\text {d }}$ | 69.2 | 67.4 | $67.4{ }^{\text {e }}$ | 67.5 |
| C-6 | $18.4{ }^{e}$ | $18.6{ }^{\text {a }}$ | $18.5{ }^{\text {c }}$ | $18.6{ }^{f}$ | $18.6^{\text {c }}$ |
| 3-OMe | $58.4{ }^{\text {f }}$ | 58.5 |  |  |  |
|  |  |  | D-Ole | D-Ole | D-Ole |
| C-1 |  |  | 101.2 | 101.4 | 101.4 |
| C-2 |  |  | 36.8 | 36.8 | 36.9 |
| C-3 |  |  | 78.9 | $79.0{ }^{\text {b }}$ | 79.1 |
| C-4 |  |  | $81.5{ }^{\text {b }}$ | $81.6{ }^{\text {d }}$ | $81.5{ }^{\text {b }}$ |
| C-5 |  |  | 72.0 | 72.1 | 72.2 |
| C-6 |  |  | $18.5{ }^{\text {c }}$ | $18.4{ }^{f}$ | $18.6{ }^{\text {c }}$ |
| 3-OMe |  |  | $56.3{ }^{\text {d }}$ | $56.4{ }^{9}$ | $56.4{ }^{\text {d }}$ |
|  | L-Cym | L-Cym | L-Cym | L-Cym | L-Cym |
| C-1 | 98.8 | 98.9 | 97.2 | 97.3 | 97.3* |
| C-2 | 32.1 | 32.2 | 32.1 | 32.2 | 32.2* |
| C-3 | 73.2 | 73.3 | 73.5 | 73.6 | 76.7* |
| C-4 | 78.9 | 78.9 | 78.9 | $79.0{ }^{\text {b }}$ | 73.4* |
| C-5 | 65.1 | 65.1 | 64.7 | 64.7 | 66.0* |
| C-6 | $18.4{ }^{\text {e }}$ | $18.5^{\text {a }}$ | $18.3^{\text {c }}$ | $18.4{ }^{f}$ | $18.5{ }^{\text {d }}$ |
| $3-\mathrm{OMe}$ | 56.7 | 56.7 | $56.8{ }^{\text {d }}$ | $56.9{ }^{9}$ | 56.9 |
|  | Glc | Glc | Glc | Glc |  |
| C-1 | 102.2* | 102.2* | 102.1* | 102.3* |  |
| C-2 | 75.1* | 75.2* | 75.2* | 75.3* |  |
| C-3 | $78.3{ }^{* / g}$ | 78.3 * ${ }^{\text {b }}$ | $71.1^{* / e}$ | $78.4{ }^{* / b}$ |  |
| C-4 | 71.7* | 71.7* | 71.7* | 73.8* |  |
| C-5 | $78.4{ }^{\prime \prime g}$ | $78.5 *$ * | 78.2 *e | $78.6{ }^{\prime \prime}$ |  |
| C-6 | 62.9 | 62.9 | 62.8 | 62.9 |  |

${ }^{a-b}$ In each column assignments may be interchangeable between Tables 1 and 2. * The chemical shifts have the longest dipole-dipole relaxation times (PRFT measurements).

Abbreviations are as follows: D-Cym, $\beta$-D-cymaropyranose; L-Cym, $\alpha$-Lcymaropyranose; D-Ole, $\beta$-D-oleandropyranose;Digit, $\beta$-D-digitoxopyranose; Glc, $\beta$-D-glucopyranose.
and their retention times are noted in Tables 3 and 4. Retention times of peaks 1 and 3 for cymarose and peaks 2 and 4 for oleandrose were clearly different between D- and L-enantiomers. The chromatogram of the sample derived from compound (8) showed one peak [1-O-(3,5-dinitrophenylcarbamoyl)-3,4-O-isopropylidene- $\alpha$-digitoxopyranose], whose retention time agreed with that of the D-enantiomer (Table 5). The analysis was performed with (6) ( 1.9 mg ), (7) ( 1.4 mg ), and (8) ( 1.1 mg ) obtained from (3) ( 17.2 mg ). It is noteworthy that the absolute configuration of compounds (6), (7), and (8) was unequivocally established with a small amount of glycoside. A protonated molecular ion peak at $m / z 325$ suggested that the residual sugar (9) was a disaccharide having a specific rotation of $-65^{\circ}$. Comparison of the ${ }^{13} \mathrm{C}$ n.m.r. spectrum $\left(\mathrm{C}_{5} \mathrm{D}_{5} \mathrm{~N}\right)$ with that of an authentic sample confirmed sugar (9) to be glaucobiose ${ }^{7}$ (4-O-$\beta$-d-glucopyranosyl-L-cymaropyranose).

Table 3. Retention times (min) of four peaks of the high-performance liquid chromatograms of carbamates derived from cymarose

| Peak | D-Cymarose | (7) from (3) | L-Cymarose |
| :---: | :---: | :---: | :---: |
| 1 | 8.8 | 8.8 | 9.8 |
| 2 | 11.2 | 11.2 | 11.2 |
| 3 | 13.0 | 13.0 | 14.4 |
| 4 | 20.0 | 20.0 | 20.0 |

Peak 1, Methyl 4-O-(3,5-dinitrophenylcarbamoyl)-x-cymaropyranoside. Peak 2. Methyl 4-O-(3,5-dinitrophenylcarbamoyl)- $\beta$-cymarofuranoside. Peak 3, Methyl 4-O-(3,5-dinitrophenylcarbamoyl)- $\beta$ cymaropyranoside. Peak 4, Methyl 4-O-(3,5-dinitrophenylcarbamoyl)-$x$-cymarofuranoside.

Table 4. Retention times ( min ) of four peaks on the high-performance liquid chromatograms of carbamates derived from oleandrose

| Peak | D-Oleandrose | (7) from (3) | (7) from (4) | L-Oleandrose |
| :---: | :---: | :---: | :---: | :---: |
| 1 | 10.7 | 10.7 | 10.7 | 11.2 |
| 2 | 14.0 | 14.0 | 14.0 | 15.3 |
| 3 | 17.1 | 17.1 | 17.1 | 17.9 |
| 4 | 25.4 | 25.4 | 25.4 | 29.6 |

Peak 1, Methyl 4-O-(3,5-dinitrophenylcarbamoyl)-x-oleandrofuranoside. Peak 2, Methyl 4-O-(3,5-dinitrophenylcarbamoyl)- $\beta$-oleandrofuranoside. Peak 3, Methyl 4-O-(3,5-dinitrophenylcarbamoyl)-x-oleandropyranoside. Peak 4, Methyl 4-O-(3,5-dinitrophenylcarbamoyl)- $\beta$ oleandropyranoside.


Figure 1. High-performance liquid chromatogram of carbamates derived from (a) authentic D-cymarose, (b) (6) in (3), (c) authentic L-cymarose, (d) authentic D-oleandrose, (e) (7) in (3), (f) (7) in (4), (g) authentic L-oleandrose, (h) authentic D-digitoxose, (i) (8) in (3), (j) (8) in (4), and (k) authentic L-digitoxose. Conditions: column, SUMIPAX OA-1000 ( $5 \mu ; 4 \mathrm{~mm}$ i.d. $\times 15 \mathrm{~cm}$ ); mobile phase, hexane-1,2-dichloroethane-ethanol (30:6:1) for cymarose, hexane-1,2-dichloroethane ethanol (40:5:2) for oleandrose, and hexane-1,2-dichloroethane-ethanol (150:3:8) for digitoxose; flow rate. $1 \mathrm{ml} \mathrm{min}^{-1}$; detector, u.v. ( 254 nm )

The terminal $\beta$-D-glucopyranosyl signals of compound (3) were easily distinguished from other sugar signals by partially relaxed Fourier-transform (PRFT) measurements ${ }^{8}$ in the ${ }^{13} \mathrm{C}$ n.m.r. spectrum (Table 2). The enzymatic hydrolysis of cyna-foside-C (3) with snail $\beta$-glucosidase gave deglucosyl-(3), compound (10). The terminal sugar of compound (10) was confirmed to be l-cymarose (11) by the presence of glaucobiose (9) in the acidic hydrolysate of compound (3). In the 500 MHz ${ }^{1}$ H n.m.r. spectrum of compound (10), one $x$-linkage and three $\beta$-linkages of four 2-deoxy sugars were revealed by the coupling constants of anomeric protons at $\delta_{\mathrm{H}} 4.51(1 \mathrm{H}, \mathrm{dd}, J 9.5$ and 1.8
$\mathrm{Hz}), 4.83(2 \mathrm{H}, \mathrm{dd}, J 10$ and 2 Hz$)$, and $4.86(1 \mathrm{H}, \mathrm{dd}, J 3$ and 1 Hz ), and all of the sugar signals were completely assigned to protons of $\alpha$ - and $\beta$-cymaropyranose, $\beta$-D-oleandropyranose, and $\beta$-d-digitoxopyranose by means of the proton decoupling measurements. The acetylation of compound (10) gave diacetate (12). Owing to the acetylation, the $3-\mathrm{H}$ signal of $\beta$-ddigitoxopyranose was shifted from $\delta_{\mathrm{H}} 4.20(1 \mathrm{H}$, ddd, $J 3,3$, and 2.9 Hz ) to $\delta_{\mathrm{H}} 5.34$ and the $4-\mathrm{H}$ signal of $\alpha$-cymaropyranose was similarly shifted from $\delta_{\mathrm{H}} 3.26(1 \mathrm{H}$, dd, $J 9.3$ and 2.9 Hz$)$ to $\delta_{\mathrm{H}} 4.41$, indicating that the $\mathrm{C}-3$ hydroxy group of the digitoxose moiety (8) was free in compound (10) and that the terminal


Figure 2. Results of n.O.e. experiments with compound (10)
sugar was $x-L$-cymaropyranose. The sugar sequence of compound (10) was suggested by the nuclear Overhauser effect (n.O.e.) difference spectra (Figure 2). Irradiation at $\delta_{\mathrm{H}} 4.86$ ( 1 H , dd, $J 3$ and $1 \mathrm{~Hz}, 1-\mathrm{H}$ of $\alpha$-L-cymaropyranose) caused enhancement at $\delta_{\mathrm{H}} 3.11(1 \mathrm{H}, \mathrm{t}, J 8.8 \mathrm{~Hz}, 4-\mathrm{H}$ of $\beta$-Doleandropyranose) as well as at $\delta_{\mathrm{H}} 1.73(1 \mathrm{H}$, ddd, $J 15.0,4.8$, and $3.7 \mathrm{~Hz}, 2-\mathrm{H}_{\mathrm{ax}}$ of $x$-L-cymaropyranose). Irradiation at $\delta_{\mathrm{H}} 3.11$ also caused enhancement at $\delta_{\mathrm{H}} 4.86$, which showed that the cymarose (11) was $x$-linked to the $\mathrm{C}-4$ hydroxy group of (7). Irradiation at $\delta_{\mathrm{H}} 3.19(1 \mathrm{H}, \mathrm{dd}, J 9.3$ and $2.9 \mathrm{~Hz}, 4-\mathrm{H}$ of $\beta-\mathrm{d}-$ digitoxopyranose) caused n.O.e. at $\delta_{\mathrm{H}} 4.51(1 \mathrm{H}, \mathrm{dd}, J 9.5$ and 1.8 $\mathrm{Hz}, 1-\mathrm{H}$ of $\beta$-D-oleandropyranose), and thus the oleandrose moiety (7) was $\beta$-linked to the $\mathrm{C}-4$ hydroxy group of digitoxose (8). Irradiation at $\delta_{\mathrm{H}} 4.83(2 \mathrm{H}$, dd, $J 10$ and $2 \mathrm{~Hz}, 1-\mathrm{H}$ of $\beta$-D-cymaropyranose and $\beta$-D-digitoxopyranose) caused n.O.e. enhancements at $\delta_{\mathrm{H}} 3.22(1 \mathrm{H}, \mathrm{dd}, J 9.3$ and $2.9 \mathrm{~Hz}, 4-\mathrm{H})$ of $\beta$-Dcymaropyranose) and $\delta_{\mathrm{H}} 3.79(1 \mathrm{H}$, ddd, $J 3,3$, and 2.9 Hz , 3 -H of $\beta$-D-cymaropyranose) as well as at $\delta_{\mathrm{H}} 3.78$ and 3.83 (each 1 H , dq, $J 9.3$ and $6.2 \mathrm{~Hz}, 5-\mathrm{H}$ of $\beta$-d-digitoxopyranose and $\beta$-D-cymaropyranose, respectively), which indicated that the digitoxose moiety (8) was $\beta$-linked to the C-4 hydroxy group of cymarose (6). In the ${ }^{13} \mathrm{C}$ n.m.r. spectrum of compound (3), the glycosidation shifts ${ }^{9}$ of the aglycone carbon signals observed at C-2 ( -2.5 p.p.m.), C-3 $(+6.1)$, and C-4 $(-4.3)$ (Table 1) suggested that the sugar moiety was linked to the C-3 hydroxy group of aglycone (5). Consequently, the structure of compound (3) was established as cynafogenin 3-O- $\beta$-D-gluco-pyranosyl-( $1 \longrightarrow 4$ )- $\alpha$-L-cymaropyranosyl- $(1 \longrightarrow 4)$ - $\beta$-D-oleandropyranosyl-( $1 \longrightarrow 4$ )- $\beta$-D-digitoxopyranosyl-( $1 \longrightarrow$ 4)- $\beta$-D-cymaropyranoside.

Compound (4) has the molecular formula $\mathrm{C}_{62} \mathrm{H}_{94} \mathrm{O}_{24}$ on the basis of elemental analysis, and field-desorption mass spectrometry (f.d.-m.s.) gave an ion peak at $m / z 1245[(M+$ $\mathrm{Na})^{+}$]. Acidic hydrolysis of compound (4) afforded an aglycone (5) and three sugars (7), (8), and (9), which were identified by comparisons of their physicochemical properties and ${ }^{1} \mathrm{H}$ or ${ }^{13} \mathrm{C}$ n.m.r. spectra. The absolute stereochemistry of compounds (7) $(1.5 \mathrm{mg})$ and ( 8 ) ( 2.8 mg ) from cynafoside-D (4) was also determined to be the D-series by the same methods as for compound (3) (Figure 1 and Tables 4 and 5). In the ${ }^{13} \mathrm{C}$ n.m.r. spectrum of compound (4) (Tables 1 and 2), however, five anomeric carbon signals at $\delta_{C} 96.0,97.2,100.3,101.2$, and 102.1 indicated that this glycoside contained five sugars. The proton decoupling measurements ( 500 MHz ) of compound (4) revealed all of the proton signals of $\alpha-L$-cymaropyranose, two $\beta$-Ddigitoxopyranoses, and $\beta$-D-oleandropyranose except for those of the methyl groups. The terminal sugar of compound (4) was confirmed to be $\beta$-D-glucopyranose by PRFT experiments. ${ }^{\overline{8}}$ The n.O.e. difference spectrum of (4) (Figure 3) showed that L cymarose (11) was $x$-linked to the C-4 hydroxy group of (7); irradiation of $\delta_{\mathrm{H}} 4.87(1 \mathrm{H}, \mathrm{dd}, J 3$ and $1 \mathrm{~Hz}, 1-\mathrm{H}$ of $x-\mathrm{L}-$ cymaropyranose) caused enhancement at $\delta_{\mathrm{H}} 3.12(1 \mathrm{H}, \mathrm{t}, J 9.3$ $\mathrm{Hz}, 4-\mathrm{H}$ of $\beta$-D-oleandropyranose), and hence the residual two molecules of digitoxose ( $\mathbf{8}$ ) were located between the aglycone (5) and the oleandrose moiety (7). Acetylation of cynafoside-D

Table 5. Retention times (min) of 1-O-(3,5-dinitrophenylcarbamoyl)-$3,4-O$-isopropylidene- $x$-digitoxopyranose on the high-performance liquid chromatograms

| D-Digitoxose | $\mathbf{( 8 )}$ from $(\mathbf{3})$ | (8) from (4) | L-Digitoxose |
| :---: | :---: | :---: | :---: |
| 13.5 | 13.5 | 13.5 | 14.3 |

(4) gave hexa-acetate (13) and caused acetylation shifts of the 3H signals of two $\beta$-D-digitoxopyranoses from $\delta_{\mathrm{H}} 4.22$ ( 2 H , ddd, $J 3,3$, and 2.9 Hz ) to $\delta_{\mathrm{H}} 5.31$ and $\delta_{\mathrm{H}} 5.37$, in addition to those of the $2-\mathrm{H}, 3-\mathrm{H}, 4-\mathrm{H}$, and $6-\mathrm{H}_{2}$ signals of $\beta$-d-glucopyranose. These shifts suggested that the oleandrose moiety (7) was linked to the C-4 hydroxy group of digitoxose (8), which was in turn linked to another digitoxose (8) moiety. The glycosidation shifts ${ }^{9}$ of the aglycone carbon signals were observed at C-2, C-3, and C-4 (Table 1), which showed that the sugar moiety was linked to the C-3 hydroxy group of aglycone (5). Therefore, the structure of compound (4) was deduced to be cynafogenin 3-O-$\beta$-d-glucopyranosyl-( $1 \longrightarrow 4$ )- $\alpha$-L-cymaropyranosyl-( $1 \longrightarrow$ 4)- $\beta$-D-oleandropyranosyl-( $1 \longrightarrow 4$ )- $\beta$-D-digitoxopyranosyl$(1 \longrightarrow 4)-\beta$-D-digitoxopyranoside.


Figure 3. Results of n.O.e. experiments with compound (4)

Conclusions.-We have already reported the technique to determine the absolute configurations of cymarose, ${ }^{4}$ oleandrose, ${ }^{5}$ and digitoxose ${ }^{6}$ by h.p.l.c. with a chiral column. This is the first report to apply the technique to the structural elucidation of glycosides containing cymarose, oleandrose, and digitoxose. It is noticeable that the technique can be performed with a small amount of glycoside [(3) (17.2 mg) and (4) (20.1 $\mathrm{mg})$ ].

Cynafoside-A (1), -B (2), and -C (3) are the second set of examples of glycosides in which a pair of optically isomeric sugars, D- and L-cymaroses, exists in the sugar chain, following the example of the wilfosides, ${ }^{10}$ isolated from Cynanchum wilfordi Hemsley.

## Experimental

M.p.s were determined on a Kofler hot-stage apparatus and are uncorrected. Optical rotations were measured in $\mathrm{CHCl}_{3}$ (except
where stated otherwise) with a JASCO DIP-4 digital polarimeter at room temperature. U.v. spectra were obtained in ethanol with a Shimadzu UV-220 spectrometer. I.r. spectra were recorded in $\mathrm{CHCl}_{3}$ on a JASCO A-102 spectrometer. ${ }^{1} \mathrm{H}$ N.m.r. spectra were recorded on JEOL GX-500 ( 500 MHz ) and FX$100(100 \mathrm{MHz})$ spectrometers in $\mathrm{CDCl}_{3}$, and ${ }^{13} \mathrm{C}$ n.m.r. spectra on JEOL GX-270 ( 67.80 MHz ) and FX-90Q (22.5 MHz ) machines in $\mathrm{C}_{5} \mathrm{D}_{5} \mathrm{~N}$ with tetramethylsilane as internal standard. Electron-impact mass spectrometry (e.i.-m.s.) was carried out with a JEOL LMS-D- 300 mass spectrometer, and f.d.-m.s. with a JEOL JMS-01SG-2 mass spectrometer. H.p.l.c. was conducted with a Waters 204 compact model, using columns of M\&S PACK C18 ( 10 mm i.d. $\times 30 \mathrm{~cm}$ ) or SUMIPAX OA-1000 ( $5 \mu ; 4 \mathrm{~mm}$ i.d. $\times 15 \mathrm{~cm}$ ) (Sumitomo Chemical Co., Ltd.). T.l.c. was performed on Merck precoated plates (Kieselgel $60 \mathrm{~F}_{254}$ ) with the following solvent systems: (A) $\mathrm{MeOH}-\mathrm{CHCl}_{3}$ (5:95), (B) $\mathrm{MeOH}-\mathrm{CHCl}_{3}$ (15:85), (C) water- $\mathrm{MeOH}-\mathrm{CHCl}_{3}[(1: 3: 12)$, lower layer $]$, and (D) acetonehexane ( $1: 1$ ). Column chromatography was carried out on Wakogel C-200 (200 mesh) or Lobar column Lichroprep R P-8 (reversed phase). The abbreviations used are as follows: cym, cymaropyranose; ole, oleandropyranose; digit, digitoxopyranose; glc, glucopyranose.

Isolation of Glycosides.-The crude glycoside (79.89 g) reported in a previous paper ${ }^{3}$ was subjected to column chromatography on silica gel with solvents of increasing polarity from $\mathrm{CHCl}_{3}$ to $\mathrm{MeOH}-\mathrm{CHCl}_{3}(1: 1)$, to separate fraction A (19.72 g), fraction B (12.17 g), and fraction C (20.04 $\mathrm{g})$. Fraction B was rechromatographed on silica gel with $\mathrm{MeOH}-\mathrm{CHCl}_{3}(7: 93)$ to separate fraction $\mathrm{B} 1[2.47 \mathrm{~g}$, a crude fraction containing compounds (1) and (3)] and fraction B2 [ 3.08 g ; a crude fraction containing compounds (2) and (4)]. Rechromatography of fraction B1 on silica gel with hexaneacetone (2:5) and $\mathrm{MeOH}-\mathrm{CHCl}_{3}$ (5:95), and further chromatography on reversed-phase gel with water- MeOH ( $15: 85$ and $20: 80$ ), afforded a crude fraction containing compound (3), which was purified by h.p.l.c. on a M\&S PACK C 18 column to give cynafoside-C ( $\mathbf{3}$ ) $(92.3 \mathrm{mg}, 0.0017 \%$ from the dried plant), $\mathrm{R}_{\mathrm{F}}(\mathrm{B}) 0.56$. Rechromatography of fraction B 2 on silica gel with hexane-acetone ( $1: 3$ and $1: 2$ ) and $\mathrm{MeOH}-$ $\mathrm{CHCl}_{3}$ (6:94 and 7:93) and, further chromatography on reversed-phase gel with water- $\mathrm{MeOH}(15: 85)$, afforded a crude fraction containing compound (4), which was purified by h.p.l.c. (M\&S PACK C18 column) to give cynafoside-D (4) $(93.1 \mathrm{mg}$, $0.0017 \%$ ), $R_{\mathrm{F}}(\mathrm{B}) 0.51$.

Cynafoside-C (3). This was obtained as an amorphous powder, m.p. $88-90^{\circ} \mathrm{C}$ (Found: C, 60.4; H, 7.95. $\mathrm{C}_{63} \mathrm{H}_{96} \mathrm{O}_{24} \cdot \mathrm{H}_{2} \mathrm{O}$ requires $\mathrm{C}, 60.3 ; \mathrm{H}, 7.9 \%) ;[x]_{\mathrm{D}}{ }^{14}+7.6^{\circ}(c 1.06) \lambda_{\text {max }} 230(\log \varepsilon$ 3.82), 278 (2.48), and $285 \mathrm{~nm}(2.46) ; v_{\text {max. }} 3420,1730,1710$, $1700,1600,1580,1490$, and $1160 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}}(500 \mathrm{MHz}) 0.95(3$ $\left.\mathrm{H}, \mathrm{s}, 19-\mathrm{H}_{3}\right), 1.15\left(3 \mathrm{H}, \mathrm{s}, 18-\mathrm{H}_{3}\right), 1.21,1.23,1.24,1.27$ (each 3 H , d, $J 6.3 \mathrm{~Hz}, 6-\mathrm{H}_{3}$ of sugars), $1.65\left(3 \mathrm{H}, \mathrm{s}, \mathrm{MeCO}_{2}\right), 2.07(3 \mathrm{H}, \mathrm{s}, 21-$ $\left.\mathrm{H}_{3}\right), 2.23\left(1 \mathrm{H}\right.$, ddd, $J 15.0,3.3$, and $1 \mathrm{~Hz}, 2-\mathrm{H}_{\mathrm{eq}}$ of $\alpha$-L-cym), 2.35 ( 1 H , ddd, $J 12.5,4.4$, and $1.5 \mathrm{~Hz}, 2-\mathrm{H}_{\mathrm{eq}}$ of $\beta$-D-ole), $3.12(1 \mathrm{H}, \mathrm{t}, J$ $8.8 \mathrm{~Hz}, 4-\mathrm{H}$ of $\beta$-D-ole), $3.20(1 \mathrm{H}, \mathrm{dd}, J 9.3$ and $2.9 \mathrm{~Hz}, 4-\mathrm{H}$ of $\beta$ -D-digit), 3.22 ( $1 \mathrm{H}, \mathrm{dd}, J 9.3$ and $2.9 \mathrm{~Hz}, 4-\mathrm{H}$ of $\beta$-d-cym), 3.26 ( 1 H , ddd, $J 11.0,8.8$, and $4.4 \mathrm{~Hz}, 3-\mathrm{H}$ of $\beta$-D-ole $), 3.26(1 \mathrm{H}, \mathrm{dd}, J$ 9.3 and $2.9 \mathrm{~Hz}, 4-\mathrm{H}$ of $x$-L-cym $), 3.32(1 \mathrm{H}, \mathrm{dq}, J 8.8$ and $6.3 \mathrm{~Hz}, 5-$ H of $\beta$-D-ole), 3.36 ( $3 \mathrm{H}, \mathrm{s}, 3$-OMe of $\beta$-D-ole), 3.37 ( $3 \mathrm{H}, \mathrm{s}, 3-$ OMe of $\alpha$-L-cym), 3.43 ( $3 \mathrm{H}, \mathrm{s}, 3$-OMe of $\beta$-D-cym), $3.57(1 \mathrm{H}, \mathrm{tt}$, $J 11.0$ and $4.9 \mathrm{~Hz}, 3-\mathrm{H}), 3.79(1 \mathrm{H}, \mathrm{dq}, J 9.3$ and $6.3 \mathrm{~Hz}, 5-\mathrm{H}$ of $\beta-$ D-digit), $3.80(1 \mathrm{H}$, ddd, $J 3,3$, and $2.9 \mathrm{~Hz}, 3-\mathrm{H}$ of $\beta$-D-cym), 3.83 $(1 \mathrm{H}, \mathrm{dq}, J 9.3$ and $6.3 \mathrm{~Hz}, 5-\mathrm{H}$ of $\beta$-d-cym $), 4.20(1 \mathrm{H}, \mathrm{ddd}, J 3,3$, and $2.9 \mathrm{~Hz}, 3-\mathrm{H}$ of $\beta$-D-digit), $4.37(1 \mathrm{H}, \mathrm{d}, J 7.8 \mathrm{~Hz}, 1-\mathrm{H}$ of $\beta$-Dglc), $4.52(1 \mathrm{H}, \mathrm{dd}, J 9.8$ and $1.5 \mathrm{~Hz}, 1-\mathrm{H}$ of $\beta$-D-ole), $4.83(2 \mathrm{H}$, dd, $J 10$ and $2 \mathrm{~Hz}, 1-\mathrm{H}$ of $\beta$-d-cym and $\beta$-d-digit), $4.86(1 \mathrm{H}, \mathrm{dd}, J$

3 and $1 \mathrm{~Hz}, 1-\mathrm{H}$ of $x$-L-cym $), 4.99(1 \mathrm{H}, \mathrm{d}, J 10.3 \mathrm{~Hz}, 12-\mathrm{H}), 5.41$ $(1 \mathrm{H}, \mathrm{t}, J 10.3 \mathrm{~Hz}, 11-\mathrm{H}), 7.49(2 \mathrm{H}, \mathrm{t}, J 7.8 \mathrm{~Hz}, m-\mathrm{ArH}), 7.62(1 \mathrm{H}$, $\mathrm{tt}, J 7.8$ and $1.5 \mathrm{~Hz}, p-\mathrm{ArH})$, and $8.05(2 \mathrm{H}, \mathrm{dd}, J 7.8$ and $1.5 \mathrm{~Hz}, o-$ $\mathrm{ArH}) ; \delta_{\mathrm{C}}(67.80 \mathrm{MHz})$ see Tables 1 and 2 .

Acidic Hydrolysis of Cynafoside-C (3).-A solution of compound (3) ( 17.2 mg ) in $\mathrm{MeOH}(15 \mathrm{ml})$ was allowed to react with $0.1 \mathrm{~m} \mathrm{H}_{2} \mathrm{SO}_{4}(5 \mathrm{ml})$ at $60^{\circ} \mathrm{C}$ for 15 min , then water ( 15 ml ) was added and the mixture was concentrated to 20 ml . The solution was kept at $60^{\circ} \mathrm{C}$ for a further 30 min , and extracted with ether $(20 \mathrm{ml})$. The ether layer was washed successively with saturated aqueous $\mathrm{NaHCO}_{3}(7 \mathrm{ml} \times 3)$ and saturated aqueous $\mathrm{NaCl}(7 \mathrm{ml}$ $\times 3$ ), and the solvent was evaporated off to give the aglycone (5) $(6.5 \mathrm{mg})$ as an amorphous powder, $R_{\mathrm{F}}(\mathrm{A}) 0.53$ and $R_{\mathrm{F}}(\mathrm{D})$ 0.58 ; m.p. $86-89^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}{ }^{16}+74.2^{\circ}(c 0.65)$; e.i.-m.s. $m / z 512$ $\left(M^{+}\right), 105\left(\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{CO}^{+}\right.$, base peak), and $43\left(\mathrm{CH}_{3} \mathrm{CO}^{+}\right) ; \delta_{\mathrm{H}}(100$ $\mathrm{MHz}) 0.98\left(3 \mathrm{H}, \mathrm{s}, 19-\mathrm{H}_{3}\right), 1.16\left(3 \mathrm{H}, \mathrm{s}, 18-\mathrm{H}_{3}\right), 1.65(3 \mathrm{H}, \mathrm{s}$, OCO-Me), $2.07\left(3 \mathrm{H}, \mathrm{s}, 21-\mathrm{H}_{3}\right), 4.95(1 \mathrm{H}, \mathrm{d}, J 9.8 \mathrm{~Hz}, 12-\mathrm{H}), 5.44$ $(1 \mathrm{H}, \mathrm{t}, J 9.8 \mathrm{~Hz}, 11-\mathrm{H}), 7.46-7.65(3 \mathrm{H}, \mathrm{m}, m$ - and $p-\mathrm{ArH})$, and 8.06 ( $2 \mathrm{H}, \mathrm{dd}, J 8.3$ and $2.0 \mathrm{~Hz}, o-\mathrm{ArH}$ ).

The aqueous layer was neutralized with saturated aqueous $\mathrm{Ba}(\mathrm{OH})_{2}$. The precipitates were filtered off and the filtrate was evaporated to give a syrup, which was chromatographed on silica gel with solvents of increasing polarity from $\mathrm{CHCl}_{3}$ to $\mathrm{MeOH}-\mathrm{CHCl}_{3}(2: 8)$ to afford separation into the monosaccharides (6) [1.9 mg, $R_{\mathrm{F}}(\mathrm{C}) 0.59$ and $\left.R_{\mathrm{F}}(\mathrm{D}) 0.42\right],(7)[1.4 \mathrm{mg}$, $R_{\mathrm{F}}(\mathrm{C}) 0.54$ and $\left.R_{\mathrm{F}}(\mathrm{D}) 0.41\right],(8)\left[1.1 \mathrm{mg}, R_{\mathrm{F}}(\mathrm{C}) 0.42\right.$ and $R_{\mathrm{F}}(\mathrm{D})$ $0.04]$, and (9) [2.2 mg, $R_{\mathrm{F}}(\mathrm{C}) 0.16$ ]. For (6), f.d.-m.s. $m / z 162$ $\left(M^{+}\right) ; \delta_{\mathrm{H}}(100 \mathrm{MHz}) 3.31,3.37,3.44$, and 3.54 (each $3 \mathrm{H}, \mathrm{s}, 3-$ OMe of $\alpha$ - and $\beta$-D-cymaropyranoses and $\alpha$-and $\beta$-D-cymarofuranoses. The intensities of the signals were in the proportions 15:10:9:7). For (7), f.d.-m.s. $m / z 162\left(M^{+}\right) ; \delta_{H}(100 \mathrm{MHz})$ for $\alpha-$ D-oleandropyranose $1.29\left(3 \mathrm{H}, \mathrm{d}, J 5.9 \mathrm{~Hz}, 6-\mathrm{H}_{3}\right), 1.50(1 \mathrm{H}$, ddd, $J 13.2,11.2$, and $\left.3 \mathrm{~Hz}, 2-\mathrm{H}_{\mathrm{ax}}\right), 2.34(1 \mathrm{H}$, ddd, $J 13.2,4.9$, and 1 $\left.\mathrm{Hz}, 2-\mathrm{H}_{\mathrm{eq}}\right), 3.15(1 \mathrm{H}, \mathrm{t}, J 9.3 \mathrm{~Hz}, 4-\mathrm{H}), 3.41(3 \mathrm{H}, \mathrm{s}, 3-\mathrm{OMe}), 3.60$ $(1 \mathrm{H}$, ddd, $J 11.2,9.3$, and $4.9 \mathrm{~Hz}, 3-\mathrm{H}), 3.94(1 \mathrm{H}, \mathrm{dq}, J 9.8$ and $5.9 \mathrm{~Hz}, 5-\mathrm{H})$, and $5.35(1 \mathrm{H}, \mathrm{dd}, J 3$ and $1 \mathrm{~Hz}, 1-\mathrm{H}) ; \delta_{\mathrm{H}}$ for $\beta$-Doleandropyranose $1.35\left(3 \mathrm{H}, \mathrm{d}, J 5.4 \mathrm{~Hz}, 6-\mathrm{H}_{3}\right)$ and $4.82(1 \mathrm{H}$, dd, $J 9.3$ and $2 \mathrm{~Hz}, 1-\mathrm{H}$ ). The signal intensities of $\alpha$ - and $\beta$-Doleandropyranoses were in the ratio $2: 1$. For (8), $[x]_{\mathrm{D}}{ }^{16}+44^{\circ}(c$ 0.10 in water); f.d.-m.s. $m / z 148\left(M^{+}\right) ; \delta_{\mathrm{H}}(100 \mathrm{MHz})$ for $\beta$-Ddigitoxopyranose $1.30\left(3 \mathrm{H}, \mathrm{d}, J 6.4 \mathrm{~Hz}, 6-\mathrm{H}_{3}\right), 1.67(1 \mathrm{H}$, ddd, $J$ $13.7,9.8$, and $\left.2.9 \mathrm{~Hz}, 2-\mathrm{H}_{\mathrm{ax}}\right), 2.17(1 \mathrm{H}, \mathrm{dt}, J 13.7$ and $2.4 \mathrm{~Hz}, 2-$ $\mathrm{H}_{\mathrm{eq}}$ ), $3.25(1 \mathrm{H}, \mathrm{dd}, J 9.8$ and $3.4 \mathrm{~Hz}, 4-\mathrm{H}), 3.78(1 \mathrm{H}, \mathrm{dq}, J 9.3$ and $6.4 \mathrm{~Hz}, 5-\mathrm{H}), 4.08$ ( 1 H , ddd, $J 3.4,2.9$, and $2.4 \mathrm{~Hz}, 3-\mathrm{H}$ ), and 5.10 ( 1 H , dd, $J 9.8$ and $2.4 \mathrm{~Hz}, 1-\mathrm{H}$ ). For $(9)$, $[x]_{\mathrm{D}}{ }^{16}-65^{\circ}$ (c 0.17 in water $\left[\right.$ lit., ${ }^{7}-72.6^{\circ}(c 0.81$ in water $\left.)\right]$; f.d.-m.s. $m / z 325[(M+$ $\left.\mathrm{H})^{+}\right] ; \delta_{\mathrm{C}}(22.5 \mathrm{MHz}$ ) for $\beta$-glaucobiose (main) 19.2, 37.4, 58.1 , $62.9,69.2,71.9,75.0,75.2,78.4$ (2 C), 79.6, 92.5, and 101.9; $\delta_{\mathrm{C}}$ for $x$-glaucobiose (minor) 18.5, 34.0, 64.7, 78.9, 91.6, and 102.4 .

Absolute Configurations of Monosaccharides (6), ${ }^{4}$ (7), ${ }^{5}$ and (8). ${ }^{\text {- }}$ A solution of compound ( 6 ) $(1.8 \mathrm{mg})$ in $\mathrm{MeOH}(0.5 \mathrm{ml})$ was allowed to react with $1 \% \mathrm{H}_{2} \mathrm{SO}_{4}-\mathrm{MeOH}(0.5 \mathrm{ml})$ at room temperature for 20 min , then water ( 1.0 ml ) was added and the reaction mixture was neutralized with saturated aqueous $\mathrm{Ba}(\mathrm{OH})_{2}$. The precipitates were filtered off and the filtrate was evaporated to give a mixture of methyl cymarosides. The product was dissolved in dry toluene $(0.25 \mathrm{ml})$ and allowed to react with 3,5 -dinitrophenyl isocyanate ( $c a .3 \mathrm{mg}$ ) in the presence of dry pyridine $(0.25 \mathrm{ml})$ at $60^{\circ} \mathrm{C}$ for 30 min , then the solvent was evaporated off to afford a mixture of carbamates. The mixture was dissolved in $\mathrm{MeOH}(1 \mathrm{ml})$ and analysed by a chiral h.p.l.c. column (SUMIPAX OA-1000) with hexane-1,2-dichloroethane-ethanol ( $30: 6: 1$ ), and the chromatogram was as depicted in Figure 1.

In the case of compound (7) ( 1.3 mg ), a mixture of carbamates was prepared by the same procedure described above and
analysed by the chiral h.p.l.c. column with hexane-dichloro-methane-ethanol ( $40: 5: 2$ ). The chromatogram was as shown in Figure 1.

A mixture of compound ( 8 ) $(1.1 \mathrm{mg}), 2,2$-dimethoxypropane $(3.0 \mathrm{mg})$, and a trace of toluene- $p$-sulphonic acid in $N, N-$ dimethylformamide ( 0.2 ml ) was stirred at room temperature for 30 min , and then poured into aqueous potassium carbonate. The product was extracted with $\mathrm{CHCl}_{3}$ and the extract washed with water, and dried $\left(\mathrm{MgSO}_{4}\right)$. Removal of the solvent under reduced pressure gave the isopropylidene derivative, which was carbamoylated by the same procedure described above. The carbamate was analysed by the chiral h.p.l.c. column with hexane-1,2-dichloroethane-ethanol (150:3:8) and the chromatogram was as shown in Figure 1.

Deglucosyl-(3), Compound (10),-A suspension ( 5 ml ) of compound (3) ( 41.3 mg ) in 0.3 m NaOAc buffer solution adjusted to pH 5.5 was treated with a suspension $(7 \mathrm{ml})$ of snail (Fruticola gainesil) $\beta$-glucosidase ( 176.3 mg ) at $37^{\circ} \mathrm{C}$ for 65 h . The products were extracted with $\mathrm{CHCl}_{3}(40 \mathrm{ml})$ and the solvent was evaporated off to give a syrup, which was chromatographed to afford deglucosyl-(3), compound (10) (14.8 mg ), as an amorphous powder, $R_{\mathrm{F}}(\mathrm{A}) 0.54$ and $R_{\mathrm{F}}(\mathrm{D}) 0.57$; m.p. $124-127^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{14}+13.5^{\circ}(c$ 1.04); f.d.-m.s. $m / z 1113[(M+$ $\left.\mathrm{K})^{+}\right], 1097\left[(M+\mathrm{Na})^{+}\right]$, and $1074\left(M^{+}\right) ; \lambda_{\text {max. }} 230(\log \varepsilon$ 4.12), 278 (3.08), and $285 \mathrm{~nm}(2.98) ; v_{\max .} 3400,1730,1710$, $1700,1600,1580,1490$, and $1160 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}}(500 \mathrm{MHz}) 0.95(3$ $\left.\mathrm{H}, \mathrm{s}, 19-\mathrm{H}_{3}\right), 1.15\left(3 \mathrm{H}, \mathrm{s}, 18-\mathrm{H}_{3}\right), 1.21,1.23,1.25,1.27($ each 3 H , $\mathrm{d}, J 6.2 \mathrm{~Hz}, 6-\mathrm{H}_{3}$ of sugar), $1.51(1 \mathrm{H}$, ddd, $J 12.5,11.0$, and 9.8 $\mathrm{Hz}, 2-\mathrm{H}_{\mathrm{ax}}$ of $\beta$-D-ole), $1.54(1 \mathrm{H}$, ddd, J $14.3,9.5$, and $2.9 \mathrm{~Hz}, 2$ $\mathrm{H}_{\text {eq }}$ of $\beta$-D-cym), 1.64 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{MeCO}_{2}$ ), $1.73(1 \mathrm{H}$, ddd, $J 15.0,4.8$, and $3.7 \mathrm{~Hz}, 2-\mathrm{H}_{\mathrm{ax}}$ of $\alpha$-L-cym $), 1.76(1 \mathrm{H}$, ddd, $J 13.7,9.3$, and 2.9 $\mathrm{Hz}, 2-\mathrm{H}_{\mathrm{ax}}$ of $\beta$-D-digit), $2.07\left(3 \mathrm{H}, \mathrm{s}, 21-\mathrm{H}_{3}\right), 2.13(1 \mathrm{H}, \mathrm{dt}, J 13.7$ and $2.4 \mathrm{~Hz}, 2-\mathrm{H}_{\text {eq }}$ of $\beta$-D-digit), $2.24(1 \mathrm{H}$, ddd, $J 14.3,3.7$, and 2.2 $\mathrm{Hz}, 2-\mathrm{H}_{\mathrm{eq}}$ of $\beta$-D-cym), $2.24(1 \mathrm{H}$, ddd, $J 15.0,3.3$, and $1.5 \mathrm{~Hz}, 2$ $\mathrm{H}_{\mathrm{eq}}$ of $\alpha-\mathrm{L}$-cym), $2.34\left(1 \mathrm{H}\right.$, ddd, $J 12.5,4.4$, and $2.2 \mathrm{~Hz}, 2-\mathrm{H}_{\mathrm{eq}}$ of $\beta$-D-ole), $3.11(1 \mathrm{H}, \mathrm{t}, J 8.8 \mathrm{~Hz}, 4-\mathrm{H}$ of $\beta$-D-ole), 3.19 ( $1 \mathrm{H}, \mathrm{dd}, J$ 9.3 and $2.9 \mathrm{~Hz}, 4-\mathrm{H}$ of $\beta$-D-digit), $3.22(1 \mathrm{H}, \mathrm{dd}, J 9.3$ and 2.9 Hz , $4-\mathrm{H}$ of $\beta$-D-cym $), 3.25(1 \mathrm{H}$, ddd, $J 11.0,8.8$, and $4.4 \mathrm{~Hz}, 3-\mathrm{H}$ of $\beta$ -D-ole), 3.26 ( 1 H , dd, $J 9.3$ and $2.9 \mathrm{~Hz}, 4-\mathrm{H}$ of $\alpha-\mathrm{L}$-cym), 3.31 ( 1 H , dq, $J 8.8$ and $6.2 \mathrm{~Hz}, 5-\mathrm{H}$ of $\beta$-D-ole), $3.37,3.38,3.43$ (each $3 \mathrm{H}, \mathrm{s}$, 3 -OMe of $\alpha$-L-cym, $\beta$-D-ole, and $\beta$-d-cym, respectively), 3.57 ( 1 $\mathrm{H}, \mathrm{tt}, J 11.0$ and $4.9 \mathrm{~Hz}, 3-\mathrm{H}), 3.59(1 \mathrm{H}$, ddd, $J 3,3$, and $2.9 \mathrm{~Hz}, 3-$ H of $\alpha$-L-cym), $3.78(1 \mathrm{H}, \mathrm{dq}, J 9.3$ and $6.2 \mathrm{~Hz}, 5-\mathrm{H}$ of $\beta$-d-digit), $3.79(1 \mathrm{H}$, ddd, $J 3,3$, and $2.9 \mathrm{~Hz}, 3-\mathrm{H}$ of $\beta$-d-cym), $3.83(1 \mathrm{H}, \mathrm{dq}$, $J 9.3$ and $6.2 \mathrm{~Hz}, 5-\mathrm{H}$ of $\beta$-D-cym), $4.07(1 \mathrm{H}, \mathrm{dq}, J 9.3$ and 6.2 Hz , $5-\mathrm{H}$ of $x$-L-cym), $4.20(1 \mathrm{H}$, ddd, $J 3,3$, and $2.9 \mathrm{~Hz}, 3-\mathrm{H}$ of $\beta$-Ddigit), 4.51 ( $1 \mathrm{H}, \mathrm{dd}, J 9.5$ and $1.8 \mathrm{~Hz}, 1-\mathrm{H}$ of $\beta$-D-ole), 4.83 ( 2 H , dd, $J 10$ and $2 \mathrm{~Hz}, 1-\mathrm{H}$ of $\beta$-d-cym and $\beta$-d-digit), $4.86(1 \mathrm{H}, \mathrm{dd}, J$ 3 and $1 \mathrm{~Hz}, 1-\mathrm{H}$ of $x-\mathrm{L}$-cym $), 4.99(1 \mathrm{H}, \mathrm{d}, J 9.9 \mathrm{~Hz}, 12-\mathrm{H}), 5.41(1$ $\mathrm{H}, \mathrm{t}, J 9.9 \mathrm{~Hz}, 11-\mathrm{H}), 7.49(2 \mathrm{H}, \mathrm{t}, J 7.7 \mathrm{~Hz}, m-\mathrm{ArH}), 7.62(1 \mathrm{H}, \mathrm{tt}$, $J 7.7$ and $1.5 \mathrm{~Hz}, p$-ArH), and $8.05(2 \mathrm{H}$, dd, $J 7.7$ and $1.5 \mathrm{~Hz}, o-$ $\mathrm{ArH}) ; \delta_{\mathrm{C}}(67.94 \mathrm{MHz})$ see Tables 1 and 2.

Di-O-acetyl-(10), Compound (12).-A mixture of compounds (10) $(2.4 \mathrm{mg})$, pyridine $(0.5 \mathrm{ml})$, and acetic anhydride $(0.4 \mathrm{ml})$ was stirred at room temperature. After 16 h , the mixture was worked up by the standard procedure to give the diacetate (12) $(2.0 \mathrm{mg}), \delta_{\mathrm{H}}(500 \mathrm{MHz}) 0.95\left(3 \mathrm{H}, \mathrm{s}, 19-\mathrm{H}_{3}\right), 1.15\left(3 \mathrm{H}, \mathrm{s}, 18-\mathrm{H}_{3}\right)$, $1.21,1.22,1.23,1.26$ (each $3 \mathrm{H}, \mathrm{d}, J 6.2 \mathrm{~Hz}, 6-\mathrm{H}_{3}$ of sugars), 1.64 $\left(3 \mathrm{H}, \mathrm{s}, \mathrm{MeCO}_{2}\right), 2.06\left(3 \mathrm{H}, \mathrm{s}, 21-\mathrm{H}_{3}\right), 2.09,2.11$ (each 3 H , s, $\mathrm{OAc}), 3.10(1 \mathrm{H}, \mathrm{t}, J 8.8 \mathrm{~Hz}, 4-\mathrm{H}$ of $\beta$-d-ole), $3.19(1 \mathrm{H}, \mathrm{dd}, J 9.3$ and $2.9 \mathrm{~Hz}, 4-\mathrm{H}$ of $\beta$-D-digit), 3.21 ( 1 H , dd, $J 9.3$ and $2.9 \mathrm{~Hz}, 4-\mathrm{H}$ of $\beta$-D-cym), 3.35, 3.36, 3.43 (each $3 \mathrm{H}, \mathrm{s}, 3$-OMe of sugar), 3.55 $(1 \mathrm{H}, \mathrm{tt}, J, 11.0$ and $4.9 \mathrm{~Hz}, 3-\mathrm{H}), 3.73(1 \mathrm{H}$, ddd, $J 3,3$, and 2.9 $\mathrm{Hz}, 3-\mathrm{H}$ of $x$-L-cym $), 3.80(1 \mathrm{H}$, ddd, $J 3,3$, and $2.9 \mathrm{~Hz}, 3-\mathrm{H}$ of $\beta$ -D-cym), 4.41 ( $1 \mathrm{H}, \mathrm{dd}, J 9.3$ and $2.9 \mathrm{~Hz}, 4-\mathrm{H}$ of $x-\mathrm{L}-\mathrm{cym}$ ), 4.45 ( 1 $\mathrm{H}, \mathrm{dd}, J 9.5$ and $1.8 \mathrm{~Hz}, 1-\mathrm{H}$ of $\beta$-D-ole), 4.83 ( $2 \mathrm{H}, \mathrm{dd}, J 10$ and 2
$\mathrm{Hz}, 1-\mathrm{H}$ of $\beta$-D-cym and $\beta$-D-digit), $4.86(1 \mathrm{H}, \mathrm{dd}, J 3$ and $1 \mathrm{~Hz}, 1-$ H of $\alpha$-L-cym $), 4.99(1 \mathrm{H}, \mathrm{d}, J 10.3 \mathrm{~Hz}, 12-\mathrm{H}), 5.34(1 \mathrm{H}, \mathrm{ddd}, J 3$, 3 , and $2.9 \mathrm{~Hz}, 3-\mathrm{H}$ of $\beta$-d-digit), $5.41(1 \mathrm{H}, \mathrm{t}, J 10.3 \mathrm{~Hz}, 11-\mathrm{H})$, $7.49(2 \mathrm{H}, \mathrm{t}, J 7.3 \mathrm{~Hz}, m-\mathrm{ArH}, 7.61(1 \mathrm{H}, \mathrm{t}, J 7.3 \mathrm{~Hz}, p-\mathrm{ArH})$, and $8.05(2 \mathrm{H}, \mathrm{d}, J 7.3 \mathrm{~Hz}, o-\mathrm{ArH})$.

Cynafoside- $D$ (4).-An amorphous powder, m.p. $153-155^{\circ} \mathrm{C}$ (Found: $\mathrm{C}, 58.7 ; \mathrm{H}, 7.6 . \mathrm{C}_{62} \mathrm{H}_{94} \mathrm{O}_{24} \cdot \frac{5}{2} \mathrm{H}_{2} \mathrm{O}$ requires $\mathrm{C}, 58.7 ; \mathrm{H}$, $7.9 \%) ;[\alpha]_{\mathrm{D}}{ }^{16}+12.8^{\circ}\left(c^{2} 1.00\right)$ f.d.-m.s. $m / z 1245\left[(M+\mathrm{Na})^{+}\right]$; $\lambda_{\text {max }} 230(\log \varepsilon 3.68), 275(2.39)$, and $282 \mathrm{~nm}(2.39) ; v_{\text {max. }} 3400$, $1730,1710,1600,1580,1490$, and $1160 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}}(500 \mathrm{MHz})$ $0.95\left(3 \mathrm{H}, \mathrm{s}, 19-\mathrm{H}_{3}\right), 1.15\left(3 \mathrm{H}, \mathrm{s}, 18-\mathrm{H}_{3}\right), 1.22,1.23,1.25,1.27$ (each $3 \mathrm{H}, \mathrm{d}, J 5.9 \mathrm{~Hz}, 6-\mathrm{H}_{3}$ of sugars), $1.51(1 \mathrm{H}$, ddd, $J 12.5$, 11.0 , and $9.8 \mathrm{~Hz}, 2-\mathrm{H}_{\mathrm{ax}}$ of $\beta$-D-ole), $1.64\left(3 \mathrm{H}, \mathrm{s}, \mathrm{MeCO}_{2}\right), 1.68$, 1.71 (each 1 H , ddd, $J 13.7,9.3$, and $2.9 \mathrm{~Hz}, 2-\mathrm{H}_{\mathrm{ax}}$ of $\beta$-D-digit), $1.79\left(1 \mathrm{H}\right.$, ddd, $J 15.0,4.8$, and $3.7 \mathrm{~Hz}, 2-\mathrm{H}_{\mathrm{ax}}$ of $\alpha-\mathrm{L}-\mathrm{cym}$ ), 2.06 ( 3 $\mathrm{H}, \mathrm{s}, 21-\mathrm{H}_{3}$ ), 2.04, 2.12 (each $1 \mathrm{H}, \mathrm{dt}, J 13.7$ and $2.4 \mathrm{~Hz}, 2-\mathrm{H}_{\text {eq }}$ of $\beta$-D-digit), $2.22\left(1 \mathrm{H}\right.$, ddd, $J 15.0,3.3$, and $1.5 \mathrm{~Hz}, 2-\mathrm{H}_{\text {eq }}$ of $\alpha-\mathrm{L}$ cym), $2.35\left(1 \mathrm{H}\right.$, ddd, $J 12.5,4.4$, and $2.2 \mathrm{~Hz}, 2-\mathrm{H}_{\mathrm{eq}}$ of $\beta$-D-ole $)$, $3.12(1 \mathrm{H}, \mathrm{t}, J 9.3 \mathrm{~Hz}, 4-\mathrm{H}$ of $\beta$-d-ole), $3.19,3.22$ (each $1 \mathrm{H}, \mathrm{d}, J 9.3$ and $2.9 \mathrm{~Hz}, 4-\mathrm{H}$ of $\beta$-D-digit), $3.26(1 \mathrm{H}$, ddd, $J 11.0,8.8$, and 4.4 $\mathrm{Hz}, 3-\mathrm{H}$ of $\beta$-d-ole), $3.31(1 \mathrm{H}, \mathrm{dq}, J 8.8$ and $5.9 \mathrm{~Hz}, 5-\mathrm{H}$ of $\beta$-dole), 3.36, 3.37 (each $3 \mathrm{H}, \mathrm{s}, 3$-OMe of $\alpha-\mathrm{L}$-cym and $\beta$-d-ole), 3.54 $(1 \mathrm{H}, \mathrm{dd}, J 9.3$ and $2.9 \mathrm{~Hz}, 4-\mathrm{H}$ of $\alpha-\mathrm{L}-\mathrm{cym}$ ), 3.73 ( 1 H , ddd, $J 3.7$, 3.3 , and $2.9 \mathrm{~Hz}, 3-\mathrm{H}$ of $\alpha-\mathrm{L}$-cym), $3.77,3.83$ (each $1 \mathrm{H}, \mathrm{dq}, J 9.3$ and $5.9 \mathrm{~Hz}, 5-\mathrm{H}$ of $\beta$-D-digit), $3.80,3.88$ (each 1 H , dd, $J 11.5$ and 4.6 Hz and $J 11.5$ and 3.3 Hz , respectively, together $6-\mathrm{H}_{2}$ of $\beta$-Dglc), $4.22(2 \mathrm{H}$, ddd, $J 3,3$, and $2.9 \mathrm{~Hz}, 3-\mathrm{H}$ of $\beta$-D-digit), 4.25 $(1 \mathrm{H}, \mathrm{dq}, J 9.3$ and $5.9 \mathrm{~Hz}, 5-\mathrm{H}$ of $\alpha-\mathrm{L}-\mathrm{cym}), 4.38(1 \mathrm{H}, \mathrm{d}, J 7.3 \mathrm{~Hz}$, $1-\mathrm{H}$ of $\beta$-D-glc), $4.52(1 \mathrm{H}$, dd, $J 9.8$ and $2 \mathrm{~Hz}, 1-\mathrm{H}$ of $\beta$-d-ole), $4.87(1 \mathrm{H}, \mathrm{dd}, J 3$ and $1 \mathrm{~Hz}, 1-\mathrm{H}$ of $\alpha$-L-cym), $4.89,4.91$ (each 1 H , dd, $J 10$ and $2 \mathrm{~Hz}, 1-\mathrm{H}$ of $\beta$-D-digit), $4.99(1 \mathrm{H}, \mathrm{d}, J 9.8 \mathrm{~Hz}, 12-\mathrm{H})$, $5.41(1 \mathrm{H}, \mathrm{t}, J 9.8 \mathrm{~Hz}, 11-\mathrm{H}), 7.49(2 \mathrm{H}, \mathrm{t}, J 7.3 \mathrm{~Hz}, m-\mathrm{ArH}), 7.62$ $(1 \mathrm{H}, \mathrm{t}, J 7.3 \mathrm{~Hz}, p-\mathrm{ArH})$, and $8.05(2 \mathrm{H}, \mathrm{d}, J 7.3 \mathrm{~Hz}, o-\mathrm{ArH}) ; \delta_{\mathrm{C}}$ ( 67.5 MHz ) see Tables 1 and 2.

Acidic Hydrolysis of Cyanofoside-D (4).--Compound (4) ( 20.1 mg ) was allowed to react by the same procedure as for (3), and the products were separated to provide aglycone (5) (6.3 mg ), and carbohydrates ( 7 ) ( 1.5 mg ), (8) ( 2.8 mg ), and (9) ( 3.0 mg ). For (5), $R_{\mathrm{F}}(\mathrm{A}) 0.53$ and $R_{\mathrm{F}}(\mathrm{D}) 0.58$; m.p. $90-92^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{17}$ $+78.1^{\circ}(c 0.63)$; e.i.-m.s. $m / z 512\left(M^{+}\right), 105\left(\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{CO}^{+}\right.$, base peak), and $43\left(\mathrm{CH}_{3} \mathrm{CO}^{+}\right) ; \delta_{\mathrm{H}}(100 \mathrm{MHz}) 0.98\left(3 \mathrm{H}, \mathrm{s}, 19-\mathrm{H}_{3}\right)$, $1.16\left(3 \mathrm{H}, \mathrm{s}, 18-\mathrm{H}_{3}\right), 1.65\left(3 \mathrm{H}, \mathrm{s}, \mathrm{MeCO}_{2}\right), 2.07\left(3 \mathrm{H}, \mathrm{s}, 21-\mathrm{H}_{3}\right)$, $4.99(1 \mathrm{H}, \mathrm{d}, J 9.8 \mathrm{~Hz}, 12-\mathrm{H}), 5.43(1 \mathrm{H}, \mathrm{t}, J 9.8 \mathrm{~Hz}, 11-\mathrm{H}), 7.48-$ $7.65(3 \mathrm{H}, \mathrm{m}, m$ - and $p-\mathrm{ArH})$, and $8.06(2 \mathrm{H}, \mathrm{dd}, J 8.3$ and 2.0 Hz , $o-\mathrm{ArH}$ ). For (7), $R_{\mathrm{F}}(\mathrm{C}) 0.54$ and $R_{\mathrm{F}}(\mathrm{D}) 0.41$; f.d.-m.s. $m / z 162$ $\left(M^{+}\right) ; \delta_{\mathrm{H}}(100 \mathrm{MHz})$ for $x$-D-oleandropyranose $1.29(3 \mathrm{H}, \mathrm{d}, J$ $\left.5.9 \mathrm{~Hz}, 6-\mathrm{H}_{3}\right), 2.38\left(1 \mathrm{H}\right.$, ddd, $J 13.2,4.9$, and $\left.1 \mathrm{~Hz}, 2-\mathrm{H}_{\mathrm{eq}}\right), 3.16(1$ $\mathrm{H}, \mathrm{t}, J 9.3 \mathrm{~Hz}, 4-\mathrm{H}), 3.41(3 \mathrm{H}, \mathrm{s}, 3-\mathrm{OMe}), 3.58(1 \mathrm{H}, \mathrm{ddd}, J 11.2$, 9.3 , and $4.9 \mathrm{~Hz}, 3-\mathrm{H}), 3.94(1 \mathrm{H}, \mathrm{dq}, J 9.3$ and $5.9 \mathrm{~Hz}, 5-\mathrm{H})$, and $5.37(1 \mathrm{H}, \mathrm{dd}, J 3$ and $1 \mathrm{~Hz}, 1-\mathrm{H})$; $\delta_{\mathrm{H}}$ for $\beta$-D-oleandropyranose $1.35\left(3 \mathrm{H}, \mathrm{d}, J 5.4 \mathrm{~Hz}, 6-\mathrm{H}_{3}\right)$ and $4.83(1 \mathrm{H}, \mathrm{dd}, J 10$ and $2 \mathrm{~Hz}, 1-$ H) $(\alpha: \beta 2: 1)$. For (8), $R_{\mathrm{F}}(\mathrm{C}) 0.42$ and $R_{\mathrm{F}}(\mathrm{D}) 0.04 ;[\alpha]_{\mathrm{D}}{ }^{17}+42^{\circ}(c$ 0.28 in water); f.d.-m.s. $m / z 148\left(M^{+}\right)$; $\delta_{\mathrm{H}}(100 \mathrm{MHz})$ for $\beta$-Ddigitoxopyranose $1.30\left(3 \mathrm{H}, \mathrm{d}, J 6.2 \mathrm{~Hz}, 6-\mathrm{H}_{3}\right), 1.66(1 \mathrm{H}$, ddd, $J$ $13.2,9.6$, and $\left.2.9 \mathrm{~Hz}, 2-\mathrm{H}_{\mathrm{ax}}\right), 2.17(1 \mathrm{H}, \mathrm{dt}, J 13.2$ and $2.2 \mathrm{~Hz}, 2-$ $\left.\mathrm{H}_{\mathrm{eq}}\right), 3.25(1 \mathrm{H}, \mathrm{dd}, J 9.6$ and $3.4 \mathrm{~Hz}, 4-\mathrm{H}), 3.78(1 \mathrm{H}, \mathrm{dq}, J 9.6$ and $6.2 \mathrm{~Hz}, 5-\mathrm{H}), 4.07(1 \mathrm{H}$, ddd, $J 3.4,2.9$, and $2.2 \mathrm{~Hz}, 3-\mathrm{H})$, and $5.10(1 \mathrm{H}, \mathrm{dd}, J 9.6$ and $2.2 \mathrm{~Hz}, 1-\mathrm{H})$. For $(9), R_{\mathrm{F}}(\mathrm{C}) 0.16 ;[\alpha]_{\mathrm{D}}{ }^{17}$ $-68^{\circ}(c 0.30$ in water $)$; f.d.-m.s. $m / z 325\left[(M+\mathrm{H})^{+}\right] ; \delta_{\mathrm{C}}(22.5$ MHz ) for $\beta$-glaucobiose (main) 19.2, 37.4, 58.1, 62.9, 69.1, 71.8, $75.0,75.2,78.5(2 \mathrm{C}), 79.6,92.5$, and 101.9. For the h.p.l.c. experiment on a chiral column (SUMIPAX OA-1000), carbohydrates (7) ( 1.5 mg ) and ( 8 ) ( 1.8 mg ) were each converted into the carbamates by the procedure described above. The chromatogram was as shown in Figure 1.

Hexa-O-acetyl-(4), Compound (13).-Acetylation of compound (4) ( 2.0 mg ) was carried out by the same procedure as for compound (10), to afford the hexa-acetate (13) ( 1.8 mg ), $\delta_{\mathrm{H}}(500$ $\mathrm{MHz}) 0.95\left(3 \mathrm{H}, \mathrm{s}, 19-\mathrm{H}_{3}\right), 1.15\left(3 \mathrm{H}, \mathrm{s}, 18-\mathrm{H}_{3}\right), 1.21,1.22,1.23$ (each $3 \mathrm{H}, \mathrm{d}, J 5.9 \mathrm{~Hz}, 6-\mathrm{H}_{3}$ of sugars), $1.64\left(3 \mathrm{H}, \mathrm{s}, \mathrm{MeCO}_{2}\right)$, $2.005,2.012,2.02,2.08,2.09$ (each $3 \mathrm{H}, \mathrm{s}, \mathrm{OAc}$ ), $2.07(6 \mathrm{H}, \mathrm{s}, 21-$ $\mathrm{H}_{3}$ and OAc ), $3.07(1 \mathrm{H}, \mathrm{t}, J 9.3 \mathrm{~Hz}, 4-\mathrm{H}$ of $\beta$-d-ole), $3.18(1 \mathrm{H}$, ddd, $J 11.0,8.8$, and $4.4 \mathrm{~Hz}, 3-\mathrm{H}$ of $\beta$-D-ole), $3.24,3.27$ (each 1 H , dd, $J 9.3$ and $2.9 \mathrm{~Hz}, 4-\mathrm{H}$ of $\beta$-D-digit), 3.33, 3.34 (each $3 \mathrm{H}, \mathrm{s}, 3-$ OMe of $x$-L-cym and $\beta$-D-ole), $3.50(1 \mathrm{H}$, dd, $J 7.8$ and $3.4 \mathrm{~Hz}, 4-$ H of $\alpha$-L-cym), 3.57 ( 1 H , ddd, $J 3.4,3$, and $3 \mathrm{~Hz}, 3$ - H of $x$-L-cym), $3.81,3.82$ (each $1 \mathrm{H}, \mathrm{dq}, J 9.3$ and $5.9 \mathrm{~Hz}, 5-\mathrm{H}$ of $\beta$-d-digit), 4.13 , 4.22 (each 1 H , dd, $J 12.2$ and 2.9 Hz and $J 12.2$ and 4.9 Hz , respectively, $6-\mathrm{H}_{2}$ of $\beta$-D-glc), $4.24(1 \mathrm{H}, \mathrm{dq}, J 9.3$ and $5.9 \mathrm{~Hz}, 5-$ H of $\alpha$-L-cym), 4.41 ( 1 H , dd, $J 10$ and $2 \mathrm{~Hz}, 1-\mathrm{H}$ of $\beta$-d-ole), 4.67 ( $1 \mathrm{H}, \mathrm{d}, J 7.0 \mathrm{~Hz}, 1-\mathrm{H}$ of $\beta$-d-glc and 1 H , dd, $J 9.8$ and $2 \mathrm{~Hz}, 1-\mathrm{H}$ of $\beta$-D-digit), $4.78(1 \mathrm{H}, \mathrm{dd}, J 9.8$ and $2 \mathrm{~Hz}, 1-\mathrm{H}$ of $\beta$-D-digit), 4.83 $(1 \mathrm{H}, \mathrm{dd}, J 3$ and $1.5 \mathrm{~Hz}, 1-\mathrm{H}$ of $\alpha$-L-cym $), 4.96(1 \mathrm{H}, \mathrm{dd}, J 9.8$ and $7.0 \mathrm{~Hz}, 2-\mathrm{H}$ of $\beta$-D-glc), $4.99(1 \mathrm{H}, \mathrm{d}, J 9.8 \mathrm{~Hz}, 12-\mathrm{H}), 5.07,5.22$ (each $1 \mathrm{H}, \mathrm{t}, J 9.8 \mathrm{~Hz}, 3$ - and $4-\mathrm{H}$ of $\beta$-d-glc), $5.31,5.37$ (each 1 H , ddd, $J 3,3$, and $2.9 \mathrm{~Hz}, 3-\mathrm{H}$ of $\beta$-D-digit), $5.41(1 \mathrm{H}, \mathrm{t}, J 9.8 \mathrm{~Hz}, 11-$ H), $7.49(2 \mathrm{H}, \mathrm{t}, J 7.8 \mathrm{~Hz}, m-\mathrm{ArH}), 7.62(1 \mathrm{H}, \mathrm{t}, J 7.8 \mathrm{~Hz}, p-\mathrm{ArH})$, and $8.05(2 \mathrm{H}, \mathrm{d}, J 7.8 \mathrm{~Hz}, o-\mathrm{ArH})$.

## Acknowledgements

We thank Mr. K. Watanabe and Dr. M. Ikura of this University for f.d.-m.s. and the $500 \mathrm{MHz}{ }^{1} \mathrm{H}$ n.m.r. spectral measurements, respectively.

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